

Unit-1
Chapter-1

ELECTROSTATICS

Electrostatics: The study of Electric forces, Electric field and electric Potential when charges are at rest is Electrostatics

Electric charge :- The force experienced by matter when it is placed near other matter is due to "Electric charge"

→ Electric charge is a scalar quantity

→ Dimensional formula → [AT]

→ SI Unit → Coulomb

→ 1 Coulomb = 3×10^9 stat Coulomb

→ Types of Charges :-

Positive : Proton

charge = $+1.6 \times 10^{-19} \text{ C}$

mass = $1.672 \times 10^{-27} \text{ kg}$

Negative - Electron

charge = $-1.6 \times 10^{-19} \text{ C}$

mass = $9.1 \times 10^{-31} \text{ kg}$

Neutral :- Neutron

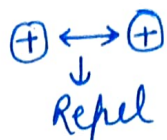
charge - Zero

Mass = $1.674 \times 10^{-27} \text{ kg}$

→ Neutron is heaviest among these.

Behaviour of Charges :

→ Similar charges always repel each other



→ Opposite charges always attract each other.



Properties :

① Additivity Property :

The total charge of a system is equal to sum of charges present in the system i.e

$$Q_{\text{Total}} = q_1 + q_2 + \dots + q_n$$

For example :-

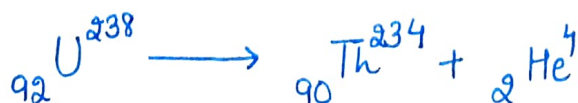
A system contains $-2q$, $-7q$, $+4q$ charge

then

$$Q_{\text{Total}} = -2q - 7q + 4q = -5q$$

② Conservation of charge :

Total charge of a system always remains constant



③ Quantisation of Charge :-

Only integral No. of charges always remain constant can be transferred from one body to another i.e fraction of charge can never be transferred.

$$Q = \pm ne$$

- Q = Total charge
- n = No. of e's
- e = Charge on e's

Ques Is a charge of $4.5 \times 10^{-19} \text{ C}$ possible?

$$Q = 4.5 \times 10^{-19} \text{ C} \quad e = 1.6 \times 10^{-19} \text{ C}$$

$$n = \frac{Q}{e} = \frac{4.5 \times 10^{-19}}{1.6 \times 10^{-19}} = 2.8$$

But according to the quantisation $n \neq 2.8$. So, charge of $4.5 \times 10^{-19} \text{ C}$ is Not Possible.

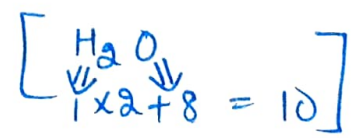
Ques Calculate the total charge and No. of e's, in a cup of 100 CC of H_2O .

Sol Molecular Weight of Water = 18; Mass of 100 cm^3 of $\text{H}_2\text{O} = 100 \text{ gm}$

$$\therefore \text{No. of molecules in } 18 \text{ gm of Water} = 6.023 \times 10^{23}$$

$$\text{No. of molecules in } 1 \text{ gm of Water} = \frac{6.023 \times 10^{23}}{18}$$

So, Each Molecule of Water contain 10 e's



$$\text{So, total 'n'} = \frac{6.023 \times 10^{23} \times 100 \times 10}{18}$$

$$n = \underline{\underline{0.33 \times 10^{26}}}$$

$$\therefore \text{Total charge} \Rightarrow Q = ne$$

$$= 0.33 \times 10^{26} \times 1.6 \times 10^{-19}$$

$$= \underline{\underline{526 \times 10^4}}$$

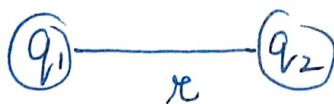
COULOMB'S LAW :-

The force of interaction b/w the charges given by

(i) directly proportional to product of charges
i.e. $F \propto q_1 q_2$

(ii) inversely proportional to sq. dis. b/w them.

$$F \propto \frac{1}{r^2}$$



$$F \propto \frac{q_1 q_2}{r^2}$$

$$F = \frac{k q_1 q_2}{r^2}$$

$$\text{where } k = \frac{1}{4\pi\epsilon_0}$$

$$\# \boxed{F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}}$$

In SI System

$$k = 9 \times 10^9 \text{ Nm}^2 \text{C}^{-2} \quad \left[k = \frac{1}{4\pi\epsilon_0} \right]$$

where

$\epsilon_0 \rightarrow$ epsilon Not
 \hookrightarrow Electrical Permittivity in vacuum.

Electrical Permittivity (ϵ_0)

It is a parameter which determines how much electric field and force affected by medium.

$$\epsilon_0 = 8.85 \times 10^{-12} \text{ N}^{-1} \text{m}^{-2} \text{C}^2$$

Relative Electrical Permittivity / Dielectric Constant (k)

The ratio of force in vacuum to the force b/w the charge in medium.

$$\frac{F_0}{F_m} = \frac{\frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}}{\frac{1}{4\pi\epsilon} \frac{q_1 q_2}{r^2}} = \frac{\frac{1}{\epsilon_0}}{\frac{1}{\epsilon}} = \frac{\epsilon}{\epsilon_0} = k$$

$$\frac{F_0}{F_m} = \frac{\epsilon}{\epsilon_0} = k$$

- \rightarrow Dielectric Constant is a unit less quantity
- $\rightarrow k = 1$ (Air)
- $\rightarrow k = \infty$ (Metal)
- $\rightarrow k = 81$ (Water)

Coulomb's law in Vector form

Let

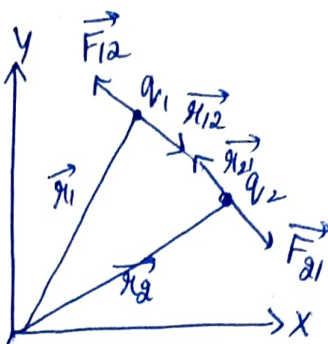
$q_1, q_2 \rightarrow$ charges placed at r_1, r_2 from origin

$$\vec{r}_{12} = \vec{r}_2 - \vec{r}_1 = \text{Dis. from } q_1 \text{ to } q_2$$

$$\vec{r}_{21} = \vec{r}_1 - \vec{r}_2 = \text{Dis. from } q_2 \text{ to } q_1$$

$$\vec{F}_{12} = \text{forces on } q_1 \text{ due to } q_2$$

$$\vec{F}_{21} = \text{forces on } q_2 \text{ due to } q_1$$



from Coulomb's law,

$$\vec{F}_{12} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{21})^2} (\hat{r}_{21}) = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{21})^3} \vec{r}_{21} \quad \text{--- (1)} \quad \left[\because \hat{r}_{21} = \frac{\vec{r}_{21}}{r_{21}} \right]$$

$$\vec{F}_{21} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{12})^2} (\hat{r}_{12}) = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{12})^3} \vec{r}_{12} \quad \text{--- (2)} \quad \left[\because \hat{r}_{12} = \frac{\vec{r}_{12}}{r_{12}} \right]$$

As

$$\boxed{\vec{r}_{12} = -\vec{r}_{21}} \quad \text{and}$$

$$\boxed{(r_{12})^3 = (r_{21})^3}$$

\therefore eqⁿ (2) becomes

$$\vec{F}_{21} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{21})^3} (-\vec{r}_{21}) \quad \text{--- (3)}$$

from eqⁿ (1) and (3)

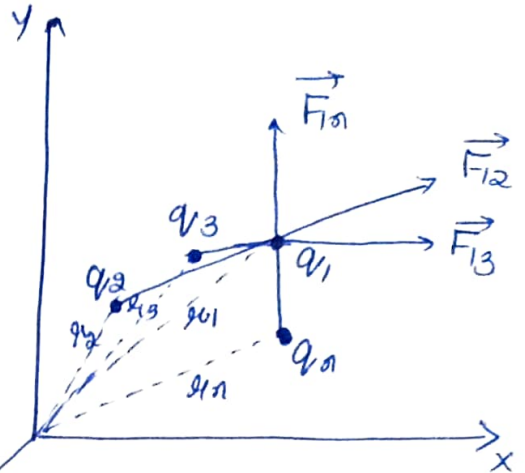
$$\boxed{\vec{F}_{12} = -\vec{F}_{21}}$$

which means action & reaction are equal and opposite. Hence Coulomb's law follows Newton's 3rd law of Motion also.

Superposition Principle for Coulomb's law ⁽⁴⁾

Let q_1, q_2, \dots, q_n no. of charges placed at r_1, r_2, \dots, r_n resp. We have to find total force on 'q' due to all charges

$$\vec{F}_{net} = \vec{F}_{12} + \vec{F}_{13} + \dots + \vec{F}_{1n}$$



$$\vec{F}_{net} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{(r_{21})^3} \vec{r}_{21} + \frac{1}{4\pi\epsilon_0} \frac{q_1 q_3}{(r_{31})^3} \vec{r}_{31} + \dots + \frac{1}{4\pi\epsilon_0} \frac{q_1 q_n}{(r_{n1})^3} \vec{r}_{n1}$$

$$\vec{F}_{net} = \frac{q_1}{4\pi\epsilon_0} \sum_{i=2}^n \frac{q_i}{(r_{i1})^3} \vec{r}_{i1}$$

Distribution of Charges

① Linear charge Distribution (λ)

The distribution of charges is over length of conductor.

$$\lambda = \frac{q}{\text{length}} = \frac{\text{Coulomb}}{\text{metre}} = \text{Cm}^{-1}$$

② Surface charge Distribution (σ)

The distribution of charges is over surface of conductor

$$\sigma = \frac{q}{\text{Area}} = \frac{\text{Coulomb}}{(\text{metre})^2} = \text{Cm}^{-2}$$

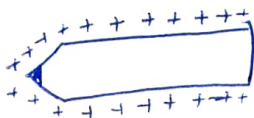
③ Volume Charge Distribution (ρ)

The distribution of charges is over volume of conductor

$$\rho = \frac{q}{\text{volume}} = \frac{\text{Coulomb}}{(\text{metre})^3} = \text{Cm}^{-3}$$

→ Smaller the length, area and volume, greater would be density of charge

→ A metallic pen is charged uniformly; maximum charge density would be at its 'tip' because it has minimum area.



Electric field : The space around a charged body in which force of interaction can be experienced by another charge is Electric field.

Source Charge : A charge which produces electric field is source charge.

Test charge :- A very small electric charge which act as 'detector' is Test charge.

Electric field Intensity (\vec{E}) : Electrostatic force per unit test charge is 'Electric field Intensity'

$$\rightarrow \text{Electric field Intensity } (\vec{E}) = \frac{\text{force } (F)}{\text{Test charge } (q_0)} = \frac{\text{Newton}}{\text{Coulomb}}$$

\rightarrow It is a vector quantity whose direction is in the direction of force.

Note :-

$$\text{Electric field Intensity } (\vec{E}) = \frac{\text{Electric Potential } (V)}{\text{length } (l)} = \text{Vm}^{-1}$$

\rightarrow Electric field Intensity is equivalent to 'Potential Gradient'

\hookrightarrow Weight of charged particle is balanced as

$$F = mg \quad \text{and} \quad F = qE$$

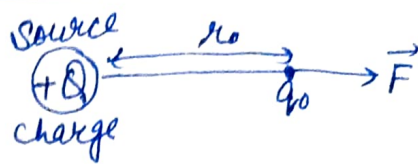
$$\boxed{mg = qE}$$

Electric Field Intensity due to a Point Charge :-

Let $Q \rightarrow$ source charge

$q_0 \rightarrow$ Test charge

$r \rightarrow$ separation b/w source + test charge



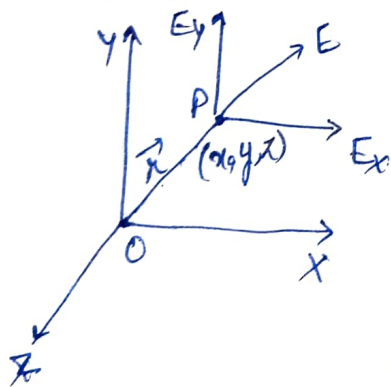
\therefore From Coulomb's law;

$$F = \frac{1}{4\pi\epsilon_0} \frac{Qq_0}{r^2}$$

$$\text{and Electric field Intensity (E)} = \frac{F}{q_0} = \frac{1}{4\pi\epsilon_0} \frac{Qq_0}{r^2} \frac{1}{q_0} = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$$

\therefore Electric field Intensity at a point depends only on source charge and separation b/w charge but not on test charge

Rectangular Components of Electric Field Intensity :-



Let $Q \rightarrow$ source charge

$$\vec{r} = x\hat{i} + y\hat{j} + z\hat{k}$$

$$r = |\vec{r}| = (x^2 + y^2 + z^2)^{1/2}$$

\therefore Electric field Intensity can be given as:

$$\vec{E} = \frac{1}{4\pi\epsilon_0} \frac{Q}{(x^2 + y^2 + z^2)^{3/2}} \hat{r}$$

$$\vec{E} = \frac{1}{4\pi\epsilon_0} \frac{Q}{(x^2 + y^2 + z^2)^{3/2}} \vec{r}$$

$$\boxed{\vec{E} = \frac{1}{4\pi\epsilon_0} \frac{Q}{(x^2 + y^2 + z^2)^{3/2}} (x\hat{i} + y\hat{j} + z\hat{k})}$$

$$\vec{E}_x = \frac{1}{4\pi\epsilon_0} \frac{Qx\hat{i}}{(x^2+y^2+z^2)^{3/2}}$$

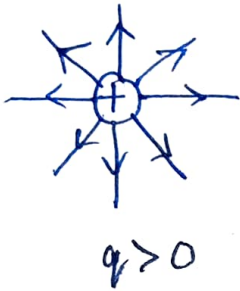
$$\vec{E}_y = \frac{1}{4\pi\epsilon_0} \frac{Qy\hat{j}}{(x^2+y^2+z^2)^{3/2}}$$

$$\vec{E}_z = \frac{1}{4\pi\epsilon_0} \frac{Qz\hat{k}}{(x^2+y^2+z^2)^{3/2}}$$

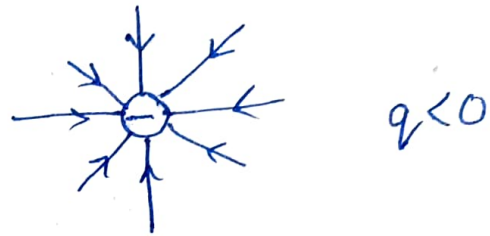
Electric field lines & Properties

Presentation of electric field is 'Electric field lines'

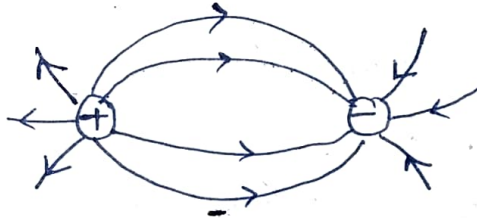
① For Positive charge



② For Negative charge



② The electric field lines of force begins from positive charge and ends on negative charge.



- ③ These field lines do not form closed loops because they travel only from +ve to -ve charge
- ④ Two electric field lines never intersect each other because if it happens then similar charge would attract each other and a single line would have two different direction which is not possible.
- ⑤ Tangent to electric field line give direction of electric field.

⑦

⑥ The electric field lines are always \perp to the conductor

⑦ The electric field intensity is more near the charge

Electric Dipole :-

A pair of equal and opposite charges separated by small distance is an "electric dipole"



Electric Dipole Moment (\vec{P}) : The product of magnitude of either charge and distance b/w them is "Electric dipole moment"

$$\boxed{\vec{P} = q \times 2a}$$

↳ SI unit of dipole moment is "Coulomb metre"

↳ CGS unit of dipole moment is "Stat Coulomb centimetre"

Polar Molecule :-

If positive and Negative charge doesnot coincide at centre is "Polar Molecule"



→ Polar Molecules have "Permanent Dipole Moment"

Non-Polar Molecule :-

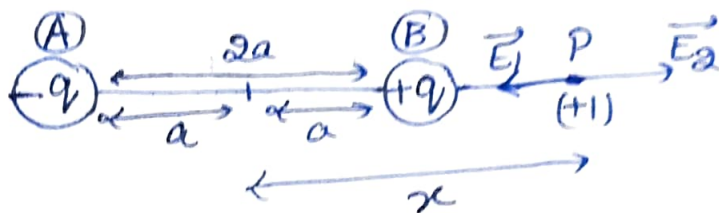
If positive and negative charges are coincide at the centre of molecules & have zero dipole moment is

Non Polar Molecule.

* Electric field Intensity at Point on Axial line of an electric dipole :

Let,

AB → An electric dipole of length ' $2a$ '



At point 'P' ; $(+1)$ test charge is placed

\vec{E}_1 → Electric field intensity b/w point P and A

\vec{E}_2 → Electric field intensity b/w point P and B

$$\therefore \boxed{E_1 = \frac{-q}{4\pi\epsilon_0(x+a)^2}}$$

$$\boxed{E_2 = \frac{+q}{4\pi\epsilon_0(x-a)^2}}$$

Net electric field Intensity at point 'P' due to both charge

$$\text{i.e. } E = E_1 + E_2$$

$$= \frac{-q}{4\pi\epsilon_0(x+a)^2} + \frac{q}{4\pi\epsilon_0(x-a)^2}$$

$$= \frac{q}{4\pi\epsilon_0} \left[\frac{-1}{(x+a)^2} + \frac{1}{(x-a)^2} \right]$$

$$\begin{aligned}
 E &= \frac{q}{4\pi\epsilon_0} \left[\frac{-(x-a)^2 + (x+a)^2}{(x^2-a^2)^2} \right] \\
 &= \frac{q}{4\pi\epsilon_0} \left[\frac{-(x^2+a^2-2xa) + (x^2+a^2+2xa)}{(x^2-a^2)^2} \right] \\
 &= \frac{q}{4\pi\epsilon_0} \left[\frac{-\cancel{x^2} - \cancel{a^2} + 2xa + \cancel{x^2} + \cancel{a^2} + 2xa}{(x^2-a^2)^2} \right] \\
 &= \frac{q}{4\pi\epsilon_0} \times \frac{4xa}{(x^2-a^2)^2}
 \end{aligned}$$

$$E = \frac{q}{4\pi\epsilon_0} \frac{2x \times 2a}{(x^2-a^2)^2} = \frac{px \cdot 2x}{4\pi\epsilon_0 (x^2-a^2)^2}$$

$$\# \boxed{E = \frac{px \cdot 2x}{4\pi\epsilon_0 (x^2-a^2)^2}} \quad \text{--- (A)}$$

Special case :-

If dipole length is very small i.e. $x \gg a$. So, a^2 can be neglected and eqⁿ (A) can be given as:

$$E = \frac{px \cdot 2x}{4\pi\epsilon_0 (x^2-0)^2} = \frac{2px}{4\pi\epsilon_0 x^4} = \frac{2p}{4\pi\epsilon_0 x^3}$$

$$\# \boxed{E = \frac{2p}{4\pi\epsilon_0 x^3}}$$